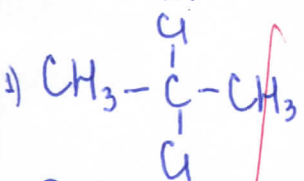
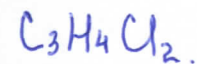


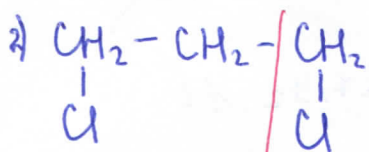
198/у 34
 % выполнения 55,88%
 работ

№ _____ от 20.09.16

№1.

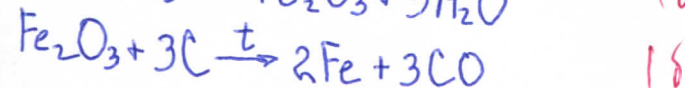
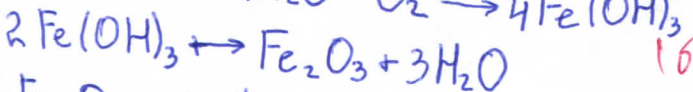
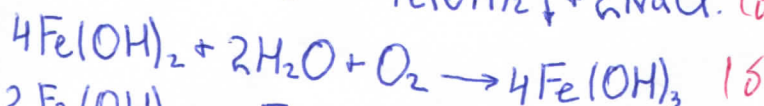
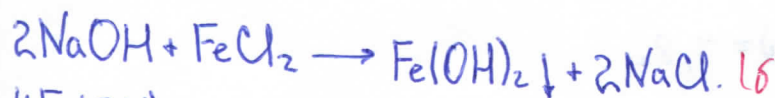


2,2 дихлорпропан 0



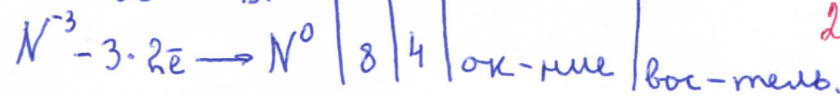
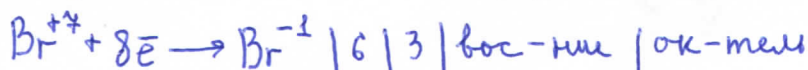
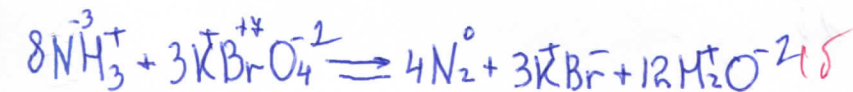
1,3 дихлорпропан. 0

№5

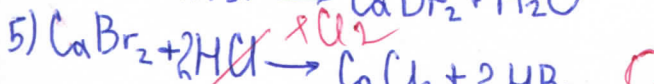
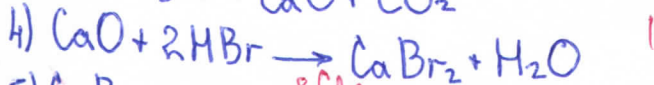
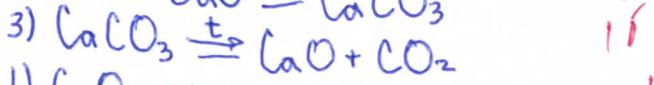
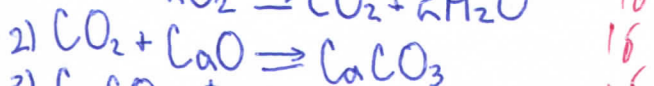
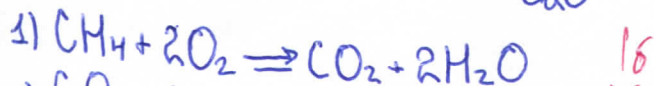
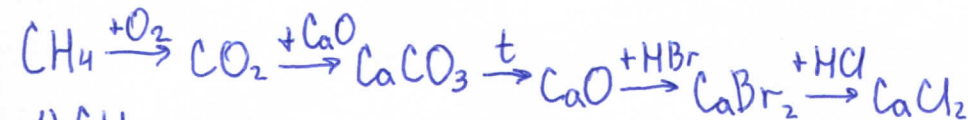


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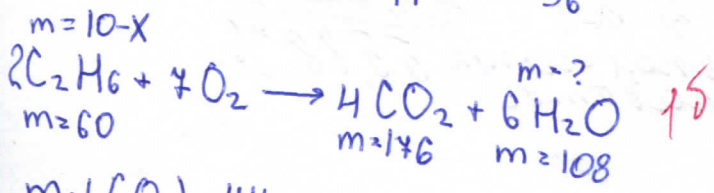
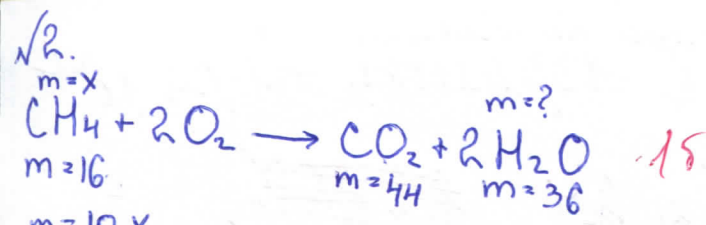
№4



№3



46



$$m_1(CO_2) = \frac{44 \cdot x}{16} = 2,75x$$

15

$$m_2(CO_2) = \frac{44(10-x)}{60} = 2,93(10-x) = 29,3 - 2,93x$$

$$m(CO_2) = 27,92 \text{ (no yacoberto).}$$

$$m(CO_2) = 2,75x + 29,3 - 2,93x = 27,92$$

15

$$29,3 - 0,18x = 27,92$$

$$-0,18x = -1,4$$

$$x = \frac{-1,4}{-0,18} = 7,8 (CH_4) \Rightarrow m(CH_4) = 7,82$$

$$m(C_2H_6) = 10 - 7,82 = 2,22$$

15

$$m_1(H_2O) = \frac{36 \cdot 7,82}{16} = 17,552$$

15

$$m_2(H_2O) = \frac{108 \cdot 2,22}{60} = 3,962$$

15

$$m(H_2O) = 17,552 + 3,962 = 21,512$$

15

Ombem: $m(H_2O) = 21,512$

85